

SYNTHETIC REAGENTS

LECTURE-5 ,DATE-07/08/2020

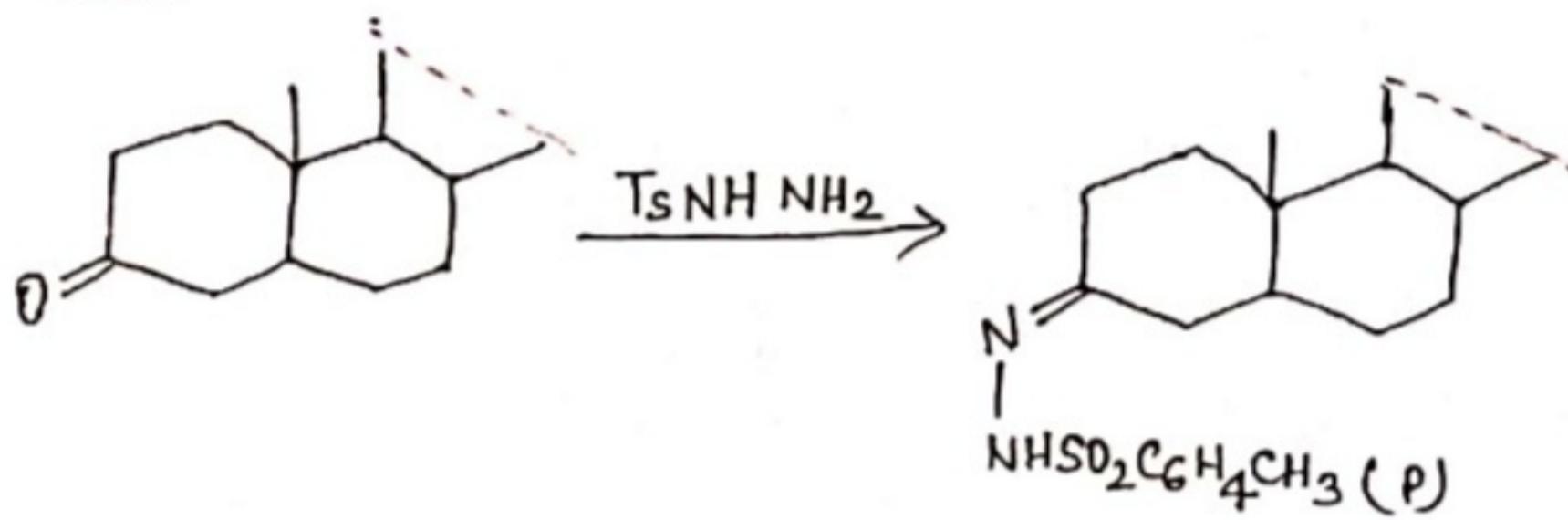
DEGREE-III (CHEMISTRY HONS.)

TOPIC :- SODIUM BOROHYDRIDE

ORGANIC CHEMISTRY , PAPER-VII

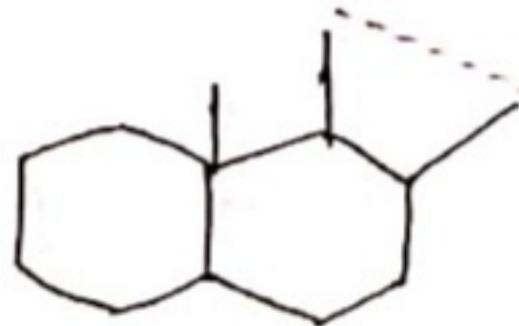
Reduction of tosylhydrazones :-

- * Tosylhydrazones can be reduced to the corresponding methylene compounds in 55-75% yield by NaBH_4 in AcOH .



Tosylhydrazone

$\downarrow \text{NaBH}_4$
 AcOH

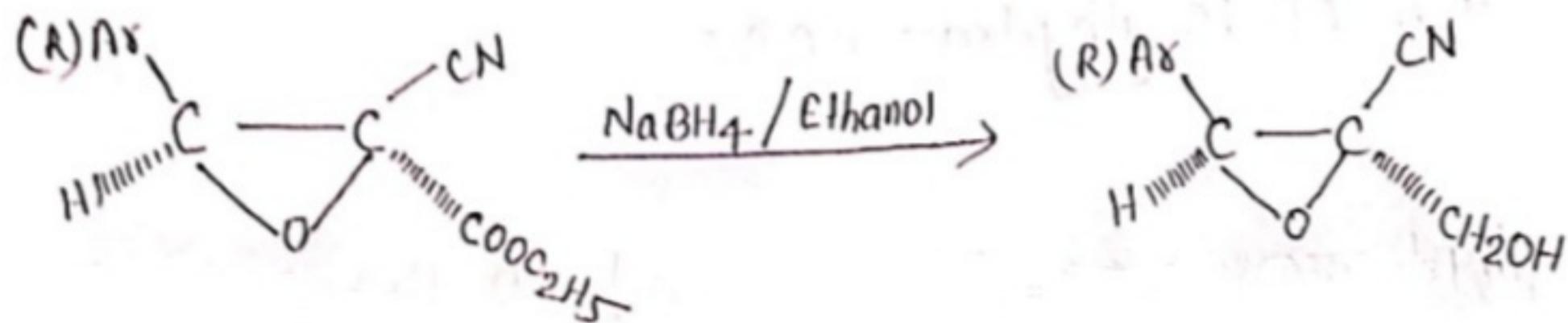


60-80 %

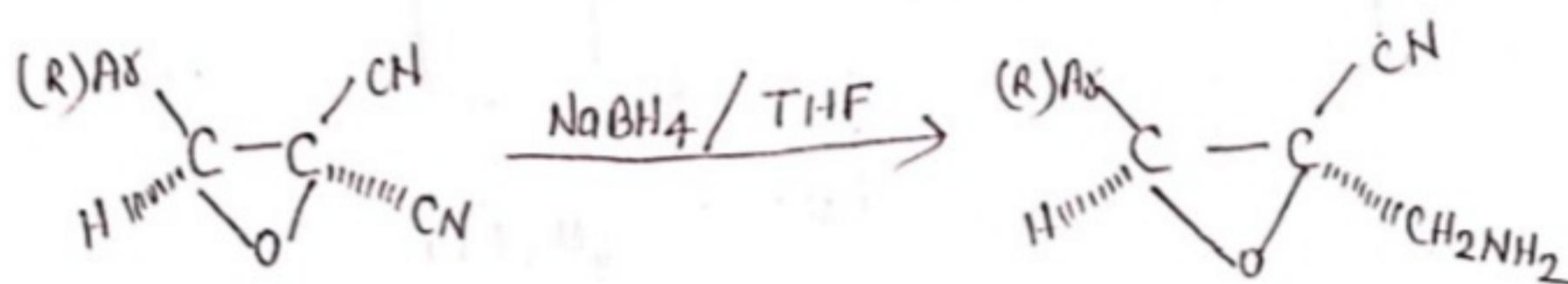
By-Dr.Rinky

Reduction of acid derivatives :-

It does not normally reduce esters, but reduction can be performed by the use of a large excess of reagent in MeOH or EtOH at room temp. or at higher temp.

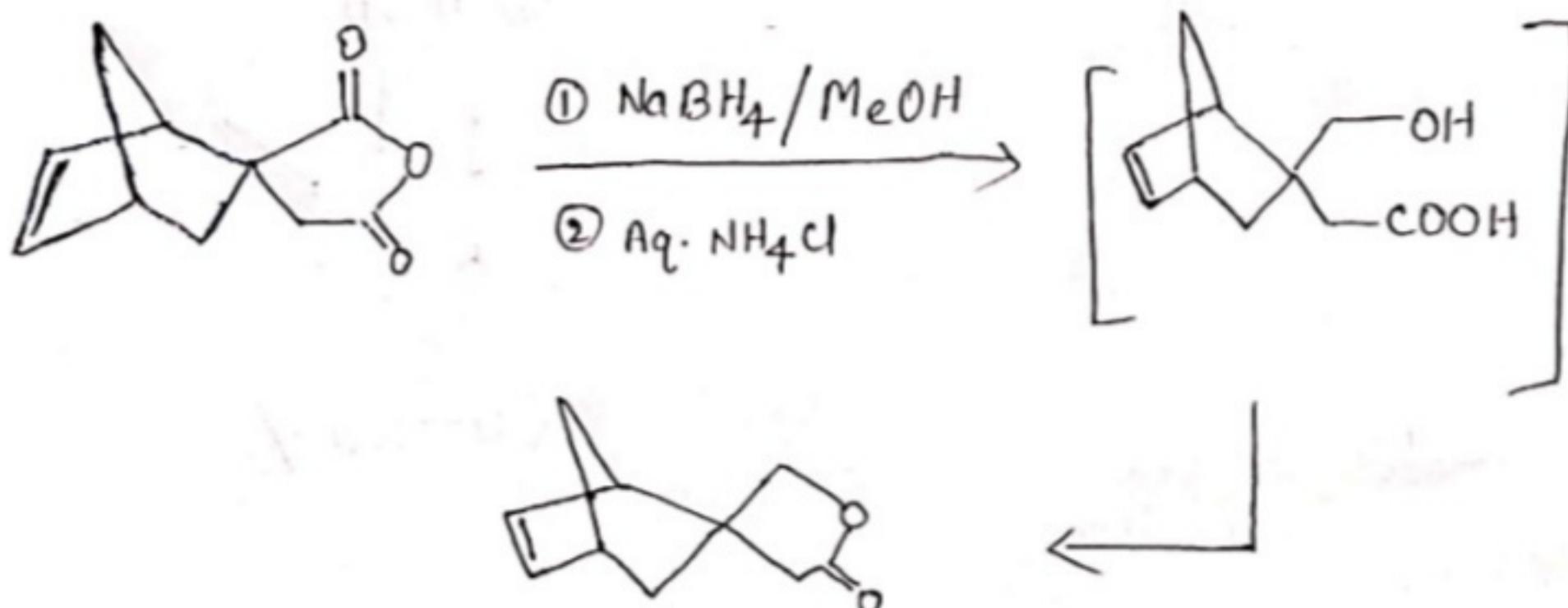


* It can reduce selectively one cyano group of 1,1-dicyano epoxide, which is trans to an alkyl or aryl substituent.

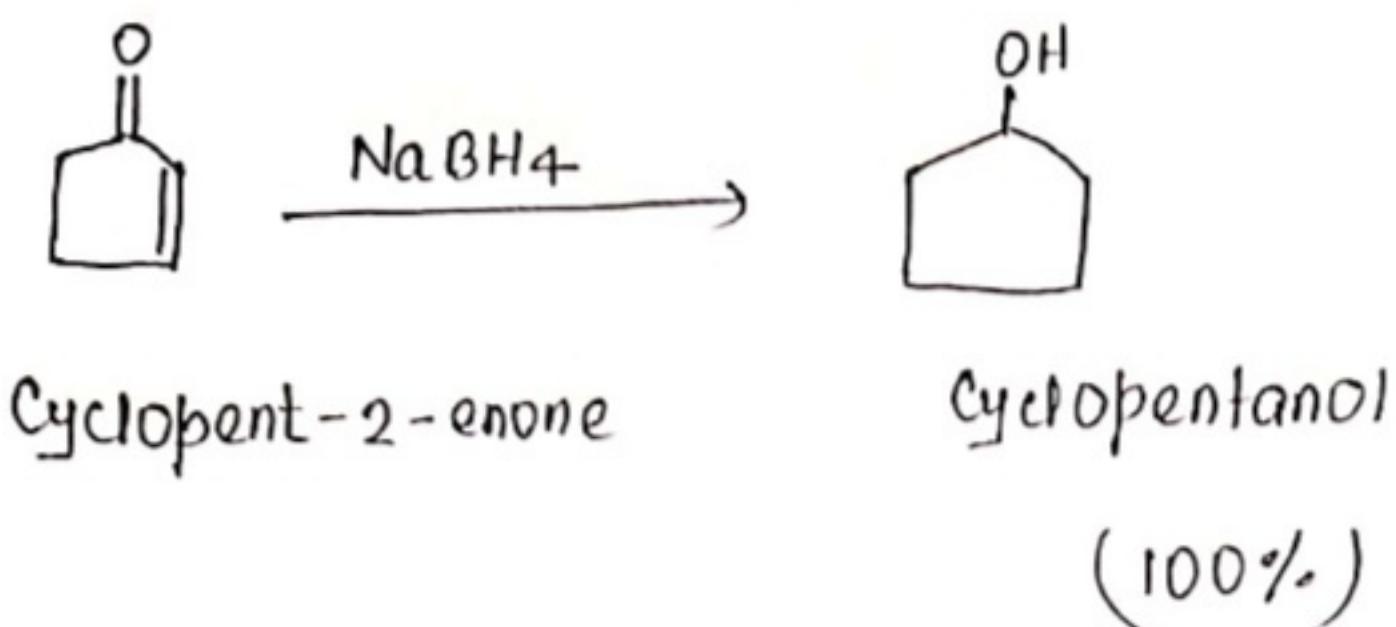


50-70 %

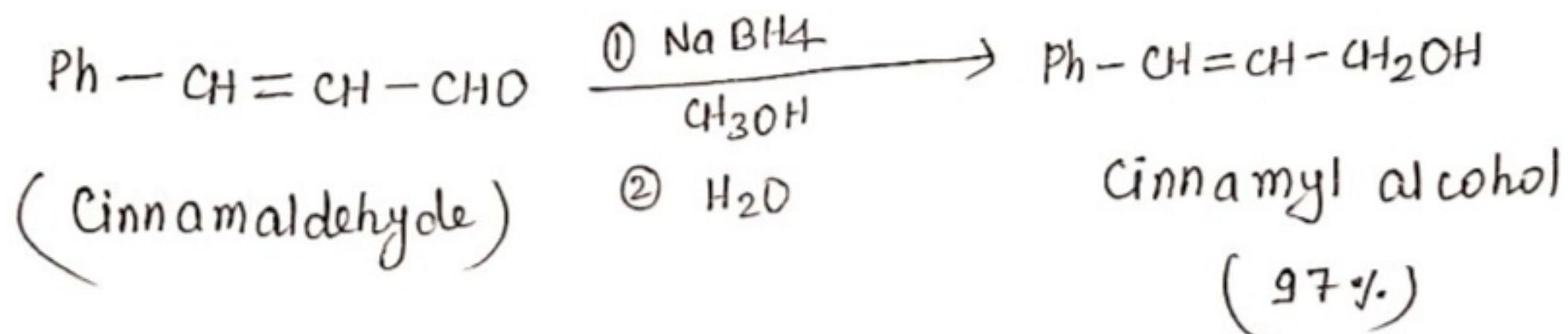
* Anhydrides may be reduced to lactones with NaBH4.



Reduction of α, β -Unsaturated carbonyls :-

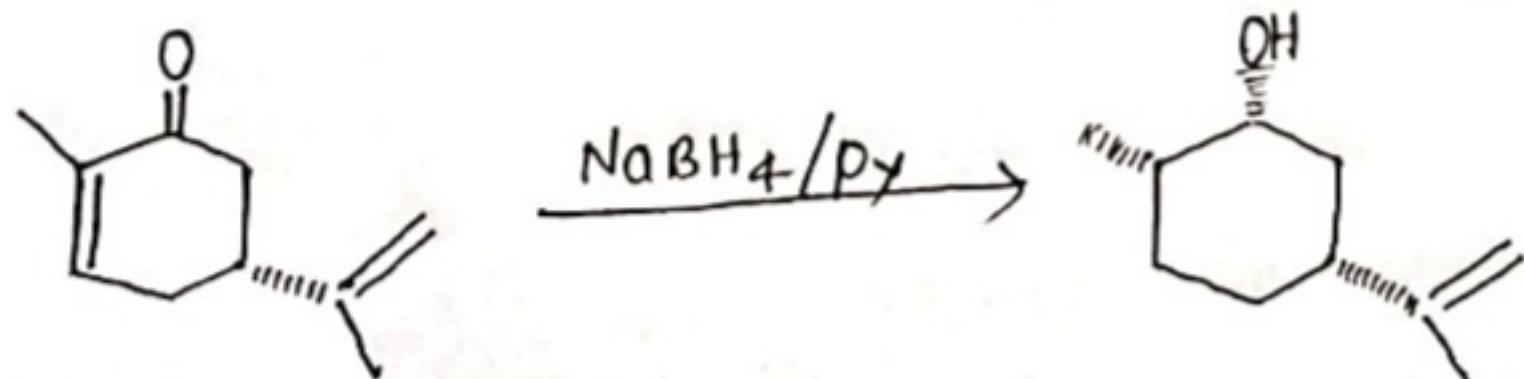


- * However, β -aryl- α,β -unsaturated aldehydes give more 1,2-reduction product than the corresponding α,β -unsaturated ketones.



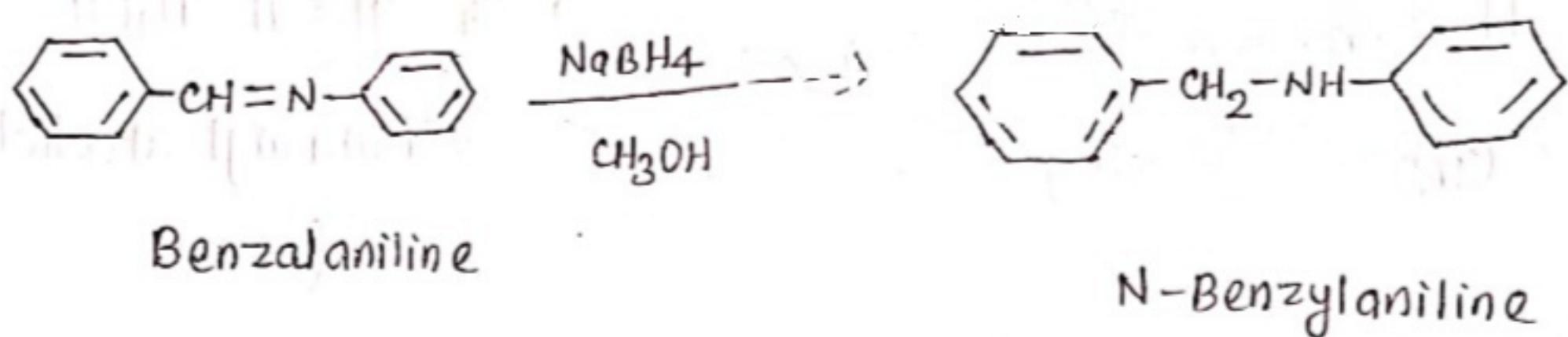
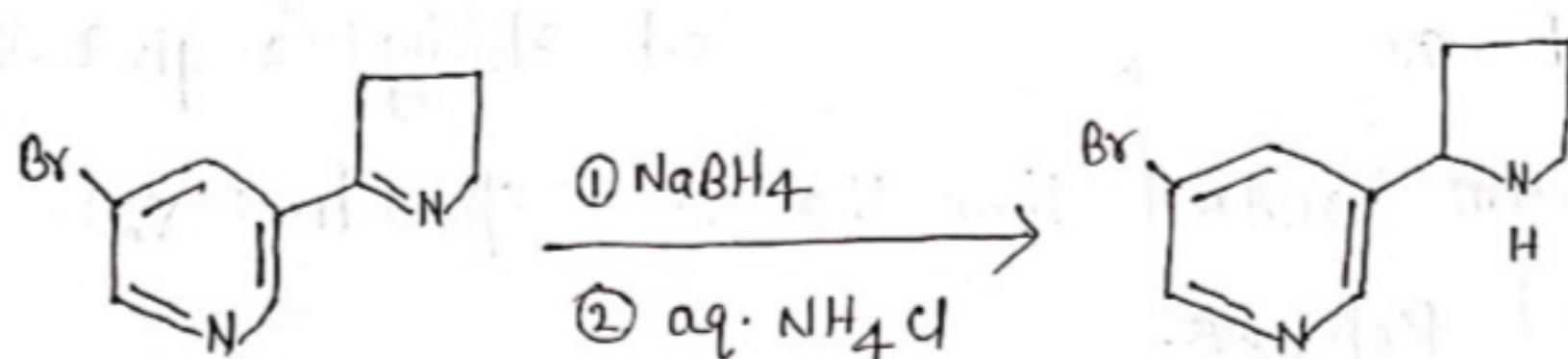
- * The reduction of adjacent double bond to carbonyl group (1,4-reduction) depends upon the conditions such as temp., solvent, amount of hydride used and the presence of catalyst.

For example: The use of pyridine as solvent increases the selectivity for 1,4-reduction as exemplified by the reduction of (R)-carvone to dihydrocarveol.



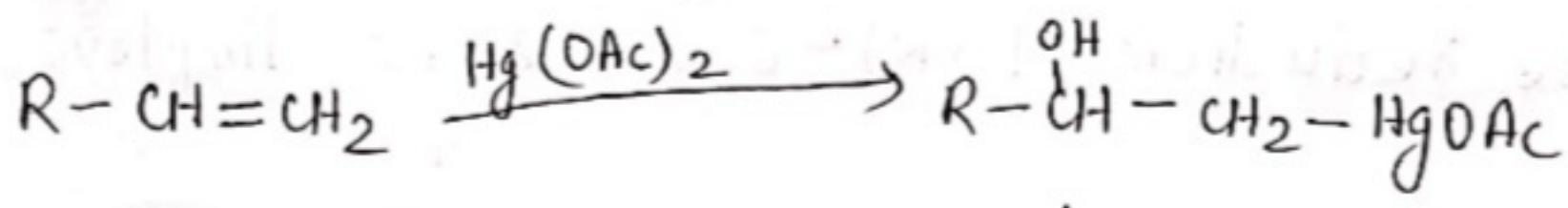
Reduction of C=N bond in enamines, imines and iminium salts.

* Imines can be reduced to amines with NaBH_4 in alcoholic solvents under neutral conditions, however, protonation of the amino nitrogen dramatically increases the rate of reduction.



NaBH_4 is used in the two-step hydration of alkene.

An alkene first treated with mercuric acetate and the process is called oxymerscuration, which is followed by reduction with NaBH_4 to generate alcohol.



SODIUM BOROHYDRIDE **COMPLETED**